# 7-endo Selective Aryl Radical Cyclization onto Enamides Leading to 3-Benzazepines: Concise Construction of a Cephalotaxine Skeleton 

Tsuyoshi Taniguchi, ${ }^{\dagger}$ Atsuko Ishita, ${ }^{\dagger}$ Masahiko Uchiyama, ${ }^{\dagger}$ Osamu Tamura, ${ }^{\dagger}$ Osamu Muraoka, ${ }^{\ddagger}$ Genzoh Tanabe, ${ }^{\ddagger}$ and Hiroyuki Ishibashi*, †
Division of Pharmaceutical Sciences, Graduate School of
Natural Science and Technology, Kanazawa University, Kanazawa 920-1192, Japan, and Faculty of Pharmaceutical Sciences, Kinki University, Higashi-osaka, Osaka 577-0818, Japan
isibasi@p.kanazawa-u.ac.jp
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$\mathrm{Bu}_{3} \mathrm{SnH}$-mediated radical cyclizations of 2-(2-bromophenyl)-$N$-ethenylacetamide gave 6 -exo cyclization product 15 as the major product, whereas $N$-[2-(2-bromophenyl)ethyl]- $N$-ethenylamides gave almost exclusively 7 -endo cyclization products. These results indicated that the position of the carbonyl group on enamide played an important role in deciding the course of the cyclization. The 7 -endo selective cyclization was applied to concise construction of a cephalotaxine skeleton.
$\mathrm{Bu}_{3} \mathrm{SnH}$-mediated cyclization of aryl radicals having a 3-butenyl group at the ortho position generally gave 5 -exo cyclization products. ${ }^{1,2}$ This was also the case for enamides 1 that gave 5-exo cyclization products, isoindolones 2 (Scheme 1). ${ }^{3}$ We recently reported, however, that enamides 3 gave only 6 -endo cyclization products, tetrahydroisoquinolines 4. ${ }^{4 \mathrm{a}}$ These results clearly indicated that the position of the carbonyl group played an important role in deciding the course of the cyclization.

[^0]SCHEME 1. Aryl Radical Cyclizations of 1 and 3


SCHEME 2. Preparations of Radical Precursors 6, 13a-d, and 14a-d


SCHEME 3. Radical Cyclizations of 6 and 13a-d


We assumed that the alkenic bond and the carbonyl group of enamides $\mathbf{1}$ and $\mathbf{3}$ were opposite each other as depicted in Scheme 1 due to thier steric and electronic repulsion. Radicals generally attack the nearest cabon atoms of the alkenic bonds, and hence enamides $\mathbf{1}$ give 5-exo cyclization products 2 and enamides $\mathbf{3}$ give 6-endo cyclization products 4.

As a continuation of our studies, we were interested in the modes of cyclization (6-exo vs 7-endo) of homologous congeners of $\mathbf{1}$ and $\mathbf{3}$, and we found that enamide $\mathbf{6}$ underwent aryl radical cyclization in a 6-exo manner to give the isoquinolinone derivative 15 and that enamides 13 and 14 gave the 7 -endo cyclization products 18 and 21 (Schemes 3 and 4), respectively. The present paper describes the results of our work in this area, including application of 7-endo selective aryl radical cyclization to concise construction of a cephalotaxine skeleton 30.

The synthesis of enamide $\mathbf{6}$ was begun by condensation of $o$-bromophenylacetic acid and $N$-ethyl-2-(phenylthio)-

## SCHEME 4. Radical Cyclizations of $14 \mathrm{a}-\mathrm{d}$


ethylamine, giving amide 5 . Treatment of 5 with $m$ chloroperbenzoic acid (MCPBA) followed by thermolysis of the resulting sulfoxide gave enamide 6 in $76 \%$ yield from 5 (Scheme 2). On the other hand, condensation of $o$-bromophenylacetic acid and 2 -(phenylthio)ethylamine followed by reduction of the resulting amide 7 with $\mathrm{BH}_{3}$ gave amine 9. Acylation of amine 9 gave amides 11, whose oxidation with MCPBA followed by thermolysis of the resulting sulfoxides afforded enamides $\mathbf{1 3 a}-\mathbf{d}$. Similarly, enamides 14a-d were prepared from o-bromophenylacetic acid and 1-methyl-2-(phenylthio)ethylamine.

When a mixture of $\mathrm{Bu}_{3} \mathrm{SnH}$ and $1,1^{\prime}$-azobis(cyclohexanecarbonitrile) (ACN) in toluene was added dropwise to a boiling solution of $\mathbf{6}$ in toluene, a 3:1 mixture of the 6 -exo cyclization product 15 and the 7 -endo cyclization product 16 was obtained in $48 \%$ combined yield along with the simple reduction product 17 ( $23 \%$ yield) (Scheme $3)$.

The cyclizations of aryl radicals having a 4-pentenyl group at the ortho position usually gave a mixture of 6 -exo and 7 -endo cyclization products. ${ }^{5}$ This was also the case for the cyclization of $\mathbf{6}$.

On the other hand, $\mathrm{Bu}_{3} \mathrm{SnH}$-mediated cyclization of enamide 13a gave the 7 -endo cyclization product 18a together with a small quantity of the 6 -exo cyclization product 19a in a ratio of 97:3 (by ${ }^{1} \mathrm{H}$ NMR) and in $42 \%$ combined yield. Enamide 13b gave only the 7 -endo cyclization product $\mathbf{1 8 b}$ in $42 \%$ yield. Similar treatment of enamide 13c gave a $98: 2$ (by ${ }^{1} \mathrm{H}$ NMR) mixture of the 7 -endo cyclization product $18 \mathbf{c}$ and the 6 -exo cyclization product 19c in $44 \%$ combined yield. Enamide 13d gave a $94: 6$ (by ${ }^{1} \mathrm{H}$ NMR) mixture of $\mathbf{1 8 d}$ and $\mathbf{1 9 d}$ in $77 \%$ combined yield. ${ }^{6,7}$ These results clearly indicated that the size of the substituent $R^{2}$ did not influence the product distribution of $\mathbf{1 8}$ and 19 . The reason the use of pivaloyl amide 13d resulted in the increase in the yield of the products, however, is obscure at the moment.

The methyl-substituted enamide 14a gave a 93:7 (by ${ }^{1} \mathrm{H}$ NMR) mixture of the 7 -endo cyclization product 21a

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FIGURE 1. Excluded pathway to the radical intermediate 27.
and the 6 -exo cyclization product 22 in $72 \%$ combined yield along with the reduction product 23a (8\%) (Scheme 4). Similar treatment of enamides $\mathbf{1 4 b} \mathbf{- d}$ gave only the 7 -endo cyclization products $\mathbf{2 1 b}-\mathbf{d}$ in $56 \%, 79 \%$ and $80 \%$ yields, respectively. No 6-exo cyclization product was detected in the crude reaction mixtures of $\mathbf{1 4 b} \mathbf{- d}$.

The high regioselectivity and the high yield of the 7 -endo cyclization products 21a-d may be due to the formation of the highly stabilized radical intermediate 27.

One possible explanation for the formation of 27 from $\mathbf{1 4 a}-\mathbf{d}$ may be a consecutive 6 -exo cyclization of aryl radicals 24 and a neophyl rearrangement of the resulting radicals 25, through the radical intermediates 26 (Figure 1). This possibility, however, was ruled out by results of the following work to simultaneously examine the effects of various $\mathrm{Bu}_{3} \mathrm{SnH}$ concentrations, addition times and reaction temperatures. ${ }^{8}$ For example, treatment of $\mathbf{1 4 b}$ ( $\mathrm{R}^{2}=\mathrm{Me}$ ) with 1.6 equiv of $\mathrm{Bu}_{3} \mathrm{SnH}$ (not using the slow addition technique) in the presence of V-70 [2, $2^{\prime}$-azobis-(2,4-dimethyl-4-methoxyvaleronitrile)] in toluene at room temperature for 10 h gave compound $\mathbf{2 1 b}$ as a sole cyclization product (see Experimental Section).

As described above, we found that the exo mode of cyclization could be shifted to the endo mode by a positional change of the carbonyl group of amides in the cyclizations of aryl radicals having a 4-pentenyl group at the ortho position.

We therefore next examined synthesis of a cephalotaxine skeleton using a radical cascade involving a 5 -endo-trig cyclization of $\alpha$-acylamino radicals such as $27 .{ }^{9}$

Cephalotaxine $(\mathbf{3 1})^{10}$ is the predominant alkaloid of Cephalotaxus species and has attracted much attention from synthetic chemists due to its unique structral features as well as the antileukemic activity of its ester derivatives, harringtonine and homoharringtonine. ${ }^{11}$
(8) Careful examinations on the effects of varying $\mathrm{Bu}_{3} \mathrm{SnH}$ concentration, $\mathrm{Bu}_{3} \mathrm{SnH}$ addition time, and reaction temperature have frequently shown that 6-endo cyclization products are formed by an initial 5-exo cyclization followed by neophyl rearrangement. See: (a) Parker, K. A.; Spero, D. M.; Inman, K. C. Tetrahedron Lett. 1986, 27, 2833. (b) Abeywickrema, A. N.; Beckwith, A. L. J.; S. Gerba, S. J. Org. Chem. 1987, 52, 4072. (c) Jones, K.; Brunton, S. A.; Gosain, R. Tetrahedron Lett. 1999, 40, 8935. See also refs 2, 3, and 5a.
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## SCHEME 5. Radial Cascade Leading to Cephalotaxine Skeleton 30



The requisite enamide 29 was readily prepared by condensation of $o$-bromophenylethylamine and cyclopentanone followed by acylation of the resulting imine 28 with acryloyl chloride (Scheme 5).

Treatment of 29 with $\mathrm{Bu}_{3} \mathrm{SnH}$ in the presence of ACN in boiling toluene gave the expected radical cascade product $\mathbf{3 0}$ in $32 \%$ yield. The structure of $\mathbf{3 0}$ was confirmed by an X-ray crystallographic analysis, and its stereochemistry was found to be identical to that of the natural cephalotaxine (31).

In conclusion, exo cyclization of aryl radicals having a 4-pentenyl group at the ortho position can be shifted to the endo mode by a positional change of the carbonyl group of enamides.

## Experimental Section

2-(2-Bromophenyl)- N -ethyl- N -(2-phenylthioethyl)acetamide (5). To a solution of (2-bromophenyl)acetic acid (3.65 g, $17.0 \mathrm{mmol})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(100 \mathrm{~mL}$ ) was added $N$-ethyl-2-(phenylthio)ethylamine ${ }^{12}$ ( $3.26 \mathrm{~g}, 14.2 \mathrm{mmol}$ ) and 1-(3-dimethylaminopro-pyl)-3-ethylcarbodiimide hydrochloride (EDC) ( $3.23 \mathrm{~g}, 16.8 \mathrm{mmol}$ ) at room temperature, and the mixture was stirred for 2 h . The reaction mixture was diluted with water and the whole was extracted with $\mathrm{CHCl}_{3}$. The extract was washed with brine, dried $\left(\mathrm{MgSO}_{4}\right)$, and concentrated. The residue was chromatographed on silica gel $\left(\mathrm{CHCl}_{3} / \mathrm{MeOH}, 100: 1\right)$ to give $5(5.72 \mathrm{~g}, 89 \%)$ as a yellow oil: $\operatorname{IR}\left(\mathrm{CHCl}_{3}\right) v 1640 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR $\left(270 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$ $\delta 1.11(33 / 100 \times 3 \mathrm{H}, \mathrm{t}, J=7.1 \mathrm{~Hz}), 1.15(67 / 100 \times 3 \mathrm{H}, \mathrm{t}, J=$ $7.1 \mathrm{~Hz}), 3.06(33 / 100 \times 2 \mathrm{H}, \mathrm{t}, J=7.4 \mathrm{~Hz}), 3.17(67 / 100 \times 2 \mathrm{H}$, $\mathrm{t}, J=7.4 \mathrm{~Hz}) 3.38(2 \mathrm{H}, \mathrm{q}, J=7.1 \mathrm{~Hz}), 3.56(2 \mathrm{H}, \mathrm{t}, J=7.4 \mathrm{~Hz})$, $3.71(33 / 100 \times 2 \mathrm{H}, \mathrm{s}), 3.77(67 / 100 \times 2 \mathrm{H}, \mathrm{s}), 7.09-7.54(9 \mathrm{H}$, $\mathrm{m})$; ${ }^{13} \mathrm{C}$ NMR ( $67.8 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 12.9,14.2,30.5,32.6,40.6$, $41.0,44.0,46.3,47.4,124.7,125.9,127.0,127.5,127.6,128.5$, $128.6,128.7,129.0,129.2,130.2,130.8,131.0,132.6,134.6,135.2$, 135.7, 169.3, 169.7. ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR spectra of 5 showed it to be a mixture of rotamers. Anal. Calcd for $\mathrm{C}_{18} \mathrm{H}_{20} \mathrm{BrNOS}$ : C, 57.14; H, 5.33; N, 3.70. Found: C, 56.85; H, 5.29; N, 3.67.

2-(2-Bromophenyl)- $\boldsymbol{N}$-ethenyl- $\boldsymbol{N}$-ethylacetamide (6). To a solution of $5(2.00 \mathrm{~g}, 5.29 \mathrm{mmol})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(100 \mathrm{~mL})$ was added dropwise a solution of MCPBA ( $1.20 \mathrm{~g}, 5.55 \mathrm{mmol}$ ) in $\mathrm{CH}_{2} \mathrm{Cl}_{2}$ $(100 \mathrm{~mL})$ at $0{ }^{\circ} \mathrm{C}$ over 30 min . To the mixture was added $10 \%$ aqueous solution of $\mathrm{Na}_{2} \mathrm{~S}_{2} \mathrm{O}_{3}$, and the mixture was stirred at the same temperature for 10 min . The mixture was washed brine, dried $\left(\mathrm{MgSO}_{4}\right)$, and concentrated. The residue was chromatographed on silica gel (hexane/AcOEt, 1:4 $\rightarrow 1: 8 \rightarrow \mathrm{AcOEt}$ ) to give 2 -(2-bromophenyl)- $N$-ethyl- $N$-(2-phenylsulfinylethyl)acetamide $(2.02 \mathrm{~g}, 97 \%)$ as an oil. A mixture of this sulfoxide $(1.00 \mathrm{~g}, 2.54$ $\mathrm{mmol})$ and $\mathrm{NaHCO}_{3}(1.07 \mathrm{~g}, 12.7 \mathrm{mmol})$ in xylene $(200 \mathrm{~mL})$ was heated at reflux for 14 h . The mixture was diluted with water and extracted with AcOEt. The organic phase was washed with brine, dried $\left(\mathrm{MgSO}_{4}\right)$, and concentrated. The residue was chromatographed on silica gel (hexane/AcOEt, $15: 1 \rightarrow 2: 1$ ) to give $\mathbf{6}$ ( $530 \mathrm{mg} \mathrm{78} \mathrm{\%}$ ) as yellow crystals: $\mathrm{mp} 59-60{ }^{\circ} \mathrm{C}$ (hexane); IR

[^2]$\left(\mathrm{CHCl}_{3}\right) v 1670,1620 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H} \operatorname{NMR}\left(270 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 1.16$ $(4 / 5 \times 3 \mathrm{H}, \mathrm{t}, J=7.1 \mathrm{~Hz}), 1.24(1 / 5 \times 3 \mathrm{H}, \mathrm{t}, J=7.1 \mathrm{~Hz}), 3.64$ $(1 / 5 \times 2 \mathrm{H}, \mathrm{q}, J=7.1 \mathrm{~Hz}), 3.75(4 / 5 \times 2 \mathrm{H}, \mathrm{q}, J=7.1 \mathrm{~Hz}) 3.93$ $(2 \mathrm{H}, \mathrm{s}), 4.37(1 \mathrm{H}, \mathrm{d}, J=9.0 \mathrm{~Hz}), 4.54(1 \mathrm{H}, \mathrm{d}, J=15.2 \mathrm{~Hz})$, $6.79(4 / 5 \mathrm{H}, \mathrm{dd}, J=15.2,9.0 \mathrm{~Hz}), 7.10-7.28(3 \mathrm{H}, \mathrm{m}), 7.44(1 / 5$ $\mathrm{H}, \mathrm{dd}, J=15.2,9.0 \mathrm{~Hz}), 7.58(1 \mathrm{H}, \mathrm{d}, J=7.9 \mathrm{~Hz}) ;{ }^{13} \mathrm{C}$ NMR $\left(67.8 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 11.8,12.8,37.1,39.0,41.1,93.7,94.2,124.7$, $127.6,128.6,130.7,131.0,132.3,132.7,134.7,168.3 .{ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR spectra of 6 showed it to be a mixture of rotamers. HRMS calcd for $\mathrm{C}_{12} \mathrm{H}_{14}{ }^{79} \mathrm{BrNO} 267.0259$, found 267.0251 .
$N$-[2-(2-Bromophenyl)ethyl]- $\boldsymbol{N}$-(2-phenylthioethyl)formamide (11a). To a stirred solution of formic acid ( 581 mg , 12.6 mmol ) in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(20 \mathrm{~mL})$ was added EDC ( $1.82 \mathrm{~g}, 9.47$ mmol ) at $0{ }^{\circ} \mathrm{C}$, and the mixture was stirred at the same temperature for 15 min . To the mixture was added a solution of $9(1.06 \mathrm{~g}, 3.16 \mathrm{mmol})$ and $N$-methylmorpholine ( $639 \mathrm{mg}, 6.31$ $\mathrm{mmol})$ in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(20 \mathrm{~mL})$, and the mixture was stirred at the same temperature for 1 h . Compound $\mathbf{9}$ was not consumed, and therefore EDC ( $605 \mathrm{mg}, 3.16 \mathrm{mmol}$ ) was added to the mixture, and the mixture was stirred for 1 h . The reaction mixture was washed with brine, dried $\left(\mathrm{MgSO}_{4}\right)$, and concentrated. The residue was chromatographed on silica gel (hexane/AcOEt, 6:1 $\rightarrow 5: 1 \rightarrow 4: 1$ ) to give 11a (1.05 g, 91\%) as a colorless oil: IR $\left(\mathrm{CHCl}_{3}\right) v 1670 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H} \operatorname{NMR}\left(270 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 2.88-3.02$ $[(2+1 / 2) \times H, m], 3.13(1 / 2 \times 2 \mathrm{H}, \mathrm{dd}, J=7.4,5.9 \mathrm{~Hz}), 3.31(1 / 2$ $\times 2 \mathrm{H}, \mathrm{t}, J=7.4 \mathrm{~Hz}), 3.45-3.57[(2 \mathrm{H}+1 / 2) \times \mathrm{H}, \mathrm{m}], 7.05-7.55$ $(9 \mathrm{H}, \mathrm{m}), 7.89(1 / 2 \mathrm{H}, \mathrm{s}), 8.01(1 / 2 \mathrm{H}, \mathrm{m}) ;{ }^{13} \mathrm{C}$ NMR ( 67.8 MHz , $\left.\mathrm{CDCl}_{3}\right) \delta 30.8,33.1,34.0,36.2,42.7,42.9,47.1,48.0,124.3,126.3$, $127.0,127.7,127.8,128.4,128.7,129.1,129.2,129.3,130.3,131.0$, $131.2,132.8,133.1,134.3,135.2,136.8,137.8,162.8,163.0 .{ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR spectra of 11a showed it to be a mixture of rotamers. Anal. Calcd for $\mathrm{C}_{17} \mathrm{H}_{18} \mathrm{BrNOS}: \mathrm{C}, 56.05$; $\mathrm{H}, 4.98$; N, 3.84. Found: C, 56.11 ; H, 5.11 ; N, 3.84 .
$\boldsymbol{N}$-[2-(2-Bromophenyl)ethyl]- N -ethenylformamide (13a). Using a procedure similar to that for the preparation of 6, compound 11a ( $858 \mathrm{mg}, 2.36 \mathrm{mmol}$ ) was treated with MCPBA ( $533 \mathrm{mg}, 2.47 \mathrm{mmol}$ ) in $\mathrm{CH}_{2} \mathrm{Cl}_{2}(160 \mathrm{~mL})$ to give $N$-[2-(2-bromophenyl)ethyl]- $N$-(2-phenylsulfinylethyl)formamide as an oil. A mixture of this sulfoxide ( $660 \mathrm{mg}, 1.74 \mathrm{mmol}$ ) and $\mathrm{NaHCO}_{3}$ ( $729 \mathrm{mg}, 8.68 \mathrm{mmol}$ ) in xylene ( 150 mL ) was heated at reflux for 14 h . After workup, the crude material was purified by chromatography on silica gel (hexane/AcOEt, 7:1 $\rightarrow 6: 1$ ) to give 13a ( $359 \mathrm{mg}, 81 \%$ ) as a colorless oil: IR $\left(\mathrm{CHCl}_{3}\right) v 1690,1635$ $\mathrm{cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR $\left(270 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 3.01(3 / 4 \times 2 \mathrm{H}, \mathrm{t}, J=7.8$ $\mathrm{Hz}), 3.08(1 / 4 \times 2 \mathrm{H}, \mathrm{t}, J=7.1 \mathrm{~Hz}), 3.74(1 / 4 \times 2 \mathrm{H}, \mathrm{t}, J=7.1$ $\mathrm{Hz}), 3.81(3 / 4 \times 2 \mathrm{H}, \mathrm{t}, J=7.8 \mathrm{~Hz}) 4.45(3 / 4 \mathrm{H}, \mathrm{dd}, J=9.2,1.7$ $\mathrm{Hz}), 4.63(1 / 4 \mathrm{H}, \mathrm{d}, J=9.2 \mathrm{~Hz}), 4.79(3 / 4 \mathrm{H}$, dd $J=15.5,1.7$ $\mathrm{Hz}), 4.80(1 / 4 \mathrm{H}, \mathrm{d}, J=16.5 \mathrm{~Hz}), 6.57(3 / 4 \mathrm{H}, \mathrm{dd}, J=15.5,9.2$ $\mathrm{Hz}), 7.07-7.30[(1 / 4+3) \mathrm{H}, \mathrm{m}], 7.53-7.59(1 \mathrm{H}, \mathrm{m}), 7.82(1 / 4$ $\mathrm{H}, \mathrm{s}), 8.31(3 / 4 \mathrm{H}, \mathrm{s}) ;{ }^{13} \mathrm{C}$ NMR ( $67.8 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 33.0,34.2$, $40.2,44.8,94.1,95.4,124.4,127.7,127.8,128.4,128.6,128.8$, $131.2,131.3,132.8,133.0,133.1,137.1,137.8,161.1,162.4 .{ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR spectra of 13a showed it to be a mixture of rotamers. Anal. Calcd for $\mathrm{C}_{11} \mathrm{H}_{12} \mathrm{BrNO}$ : C, $51.99 ; \mathrm{H}, 4.76 ; \mathrm{N}$, 5.51. Found: C, 52.07; H, 4.81; N, 5.43 .

2-Ethyl-1,2,3,4-tetrahydro-1-methylisoquinolin-2-one (15), 3-Ethyl-2,3,4,5-tetrahydro-1H-3-benzazepin-2-one (16), and $N$-Ethenyl- $N$-ethyl-2-phenylacetamide (17). General Procedure for Radical Cyclization. To a boiling solution of $\mathbf{6}$ (250 $\mathrm{mg}, 0.932 \mathrm{mmol}$ ) in toluene ( 50 mL ) was added dropwise a solution of $\mathrm{Bu}_{3} \mathrm{SnH}(407 \mathrm{mg}, 1.40 \mathrm{mmol})$ and ACN ( 45.6 mg , 0.187 mmol ) in toluene ( 50 mL ) over 1.5 h , and the mixture was further heated at reflux for 1 h . The solvent was evaporated off, $\mathrm{Et}_{2} \mathrm{O}(50 \mathrm{~mL})$ and an $8 \%$ aqueous KF solution ( 50 mL ) were added to the residue, and mixture was vigorously stirred at room temperature overnight. The organic phase was separated, and the aqueous phase was further extracted with $\mathrm{Et}_{2} \mathrm{O}$. The combined organic phase was washed with brine, dried $\left(\mathrm{MgSO}_{4}\right)$, and concentrated. The residue was chromatographed on silica gel (hexane $\rightarrow$ hexane/AcOEt, $10: 1 \rightarrow 5: 1 \rightarrow 3: 1$ ). The first fraction gave $17(40.6 \mathrm{mg}, 23 \%)$ as a colorless oil: $\mathrm{IR}\left(\mathrm{CHCl}_{3}\right) v$ $1665,1620 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H} \operatorname{NMR}\left(270 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 1.13(3 \mathrm{H}, \mathrm{t}, J=$ 7.1 Hz ), $3.60-3.76$ (total $2 \mathrm{H}, \mathrm{q}, ~ J=7.1 \mathrm{~Hz}$ ), $3.83(2 \mathrm{H}, \mathrm{s}), 4.30$ $(1 \mathrm{H}, \mathrm{d}, J=9.2 \mathrm{~Hz}), 4.48(1 \mathrm{H}, \mathrm{d}, J=15.5 \mathrm{~Hz}), 6.80(1 \mathrm{H}, \mathrm{dd}, J$
$=15.5,9.2 \mathrm{~Hz}), 7.15-7.49(5 \mathrm{H}, \mathrm{m}) ;{ }^{13} \mathrm{C} \operatorname{NMR}\left(67.8 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$ $\delta 11.7,36.9,41.1,93.9,126.9,128.7,132.6,169.3$. HRMS calcd for $\mathrm{C}_{12} \mathrm{H}_{15} \mathrm{NO} 189.1154$, found 189.1152 . The second fraction gave a mixture of 15 and $16(84.6 \mathrm{mg}, 48 \%)$ in a ratio of ca. 3:1: IR $\left(\mathrm{CHCl}_{3}\right) v 1640 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H} \operatorname{NMR}\left(500 \mathrm{MHz}, \mathrm{C}_{6} \mathrm{D}_{6}\right) \delta 0.85(1 / 4 \times$ $3 \mathrm{H}, \mathrm{t}, J=7.1 \mathrm{~Hz}$, for $\mathbf{1 6}), 0.92(3 / 4 \times 3 \mathrm{H}, \mathrm{d}, J=6.8 \mathrm{~Hz}$, for 15), $0.96(3 / 4 \times 3 \mathrm{H}, \mathrm{t}, J=7.3 \mathrm{~Hz}$, for 15$), 2.48(1 / 4 \times 2 \mathrm{H}, \mathrm{t}, J$ $=5.9 \mathrm{~Hz}$, for $\mathbf{1 6}) 2.75(3 / 4 \mathrm{H}, \mathrm{dq}, J=13.8,6.9 \mathrm{~Hz}$, for $\mathbf{1 5}), 2.86$ $(1 / 4 \times 2 \mathrm{H}, \mathrm{t}, J=9.2 \mathrm{~Hz}$, for 16$), 3.20(1 / 4 \times 2 \mathrm{H}, \mathrm{q} J=7.1 \mathrm{~Hz}$, for 16), $3.37(3 / 4 \mathrm{H}, \mathrm{d}, J=19.0 \mathrm{~Hz}$, for 15$)$; $3.52(3 / 4 \mathrm{H}, \mathrm{d}, J=$ 19.0 Hz , for 15$)$, $3.66(1 / 4 \times 2 \mathrm{H}$, s, for $\mathbf{1 6}), 3.84(3 / 4 \mathrm{H}, \mathrm{dq}, J=$ $13.9,7.0 \mathrm{~Hz}$, for 15$), 3.90(3 / 4 \mathrm{H}, \mathrm{q}, ~ J=6.7 \mathrm{~Hz}$, for 15$), 6.71-$ $7.32(3 / 4 \times 4 \mathrm{H}+1 / 4 \times 4 \mathrm{H}, \mathrm{m}$, for 15 and 16). HRMS calcd for $\mathrm{C}_{12} \mathrm{H}_{15} \mathrm{NO}$ 189.1154, found 189.1155. Anal. Calcd for $\mathrm{C}_{12} \mathrm{H}_{15}-$ NO: C, 76.16; H, 7.99; N, 7.40. Found: C, 75.75; H, 7.96; N, 7.36 .

3-Formyl-2,3,4,5-tetrahydro-1H-3-benzazepine(18a),2-Form-yl-1,2,3,4-tetrahydro-1-methylisoquinoline (19a), and $N$ -Ethenyl- $\boldsymbol{N}$-(2-phenylethyl)formamide (20a). Following the general procedure, a boiling solution of $\mathbf{1 3 a}(100 \mathrm{mg}, 0.394 \mathrm{mmol})$ in toluene $(20 \mathrm{~mL})$ was treated with $\mathrm{Bu}_{3} \mathrm{SnH}(172 \mathrm{mg}, 0.590$ mmol ) and ACN ( $19.2 \mathrm{mg}, 0.0787 \mathrm{mmol}$ ) in toluene ( 20 mL ). After workup, the crude material was chromatographed on silica gel (hexane/AcOEt, $6: 1 \rightarrow 2: 1 \rightarrow 3: 2$ ). The first fraction gave 20a ( $11.1 \mathrm{mg}, 16 \%$ ) as a colorless oil: IR $\left(\mathrm{CHCl}_{3}\right) v 1690,1635 \mathrm{~cm}^{-1}$; ${ }^{1} \mathrm{H} \operatorname{NMR}\left(270 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 2.86(3 / 4 \times 2 \mathrm{H}, \mathrm{t}, J=7.6 \mathrm{~Hz})$, $2.92(1 / 4 \times 2 \mathrm{H}, \mathrm{t}, J=6.9 \mathrm{~Hz}), 3.69(1 / 4 \times 2 \mathrm{H}, \mathrm{t}, J=6.9 \mathrm{~Hz})$, $3.79(3 / 4 \times 2 \mathrm{H}, \mathrm{t}, J=7.6 \mathrm{~Hz}) 4.47(1 \mathrm{H}, \mathrm{dd}, J=9.2,1.5 \mathrm{~Hz})$, $4.64(3 / 4 \mathrm{H}$, dd, $J=16.2,1.5 \mathrm{~Hz}), 4.67(1 / 4 \mathrm{H}$, dd $J=16.2,1.5$ $\mathrm{Hz}), 6.57(3 / 4 \mathrm{H}, \mathrm{dd}, J=16.2,9.2 \mathrm{~Hz}), 7.12-7.37[(1 / 4+5 \mathrm{H})$, $\mathrm{m}], 7.74(1 / 4 \mathrm{H}, \mathrm{s}), 8.29(3 / 4 \mathrm{H}, \mathrm{s}) ;{ }^{13} \mathrm{C}$ NMR $\left(67.8 \mathrm{MHz}, \mathrm{CDCl}_{3}\right)$ $\delta 32.5,33.6,41.8,47.0,93.8,95.3,126.6,127.0,128.6,161.1$, 162.3. ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR spectra of $20 a$ showd it to be a mixture of rotamers. HRMS calcd for $\mathrm{C}_{11} \mathrm{H}_{13} \mathrm{NO} 175.0997$, found 175.0991. The second fraction gave a mixture of $\mathbf{1 8 a}$ and $\mathbf{1 9 a}(29.0 \mathrm{mg}$, $42 \%$ ) in a ratio of 97:3: IR $\left(\mathrm{CHCl}_{3}\right) v 1670,1660 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR $\left(500 \mathrm{MHz}, \mathrm{C}_{6} \mathrm{D}_{6}\right) \delta 0.94(3 / 100 \times 3 \mathrm{H}, \mathrm{d}, J=6.8 \mathrm{~Hz}$, for $\mathbf{1 9 a}$ ), $1.97(97 / 100 \times 2 \mathrm{H}, \mathrm{t}, J=5.1 \mathrm{~Hz}$, for $\mathbf{1 8 a}), 2.13(97 / 100 \times 2 \mathrm{H}$, $\mathrm{t}, J=5.1 \mathrm{~Hz}$, for $18 \mathbf{a}), 2.33(97 / 100 \times 2 \mathrm{H}, \mathrm{t}, J=5.0 \mathrm{~Hz}$, for 18a), $3.14(97 / 100 \times 2 \mathrm{H}, \mathrm{t}, ~ J=5.0 \mathrm{~Hz}$, for $\mathbf{1 8 a}), 4.11(3 / 100 \mathrm{H}$, dd, $J=13.2,6.5 \mathrm{~Hz}$, for $\mathbf{1 9 a}), 5.24(3 / 100 \mathrm{H}, \mathrm{q}, ~ J=6.8 \mathrm{~Hz}$, for 19a), $6.43-6.60(97 / 100 \times 2 \mathrm{H}+3 / 100 \times 2 \mathrm{H}, \mathrm{m}$, for $18 \mathbf{a}$ and 19a), $6.65-6.70(97 / 100 \times 2 \mathrm{H}+3 / 100 \times 2 \mathrm{H}$, m, for $18 \mathbf{a}$ and 19a), 7.52 ( $97 / 100 \mathrm{H}$, s, for 18a), $7.72(3 / 100 \mathrm{H}$, s, for $19 \mathbf{1 9}){ }^{13}{ }^{13} \mathrm{C}$ NMR ( $125 \mathrm{MHz}, \mathrm{C}_{6} \mathrm{D}_{6}$ for 18a) $\delta 37.7,39.1,42.6,48.5,127.0$, $127.3,130.0,130.2,140.8,141.7,161.5$. HRMS calcd for $\mathrm{C}_{11} \mathrm{H}_{13}{ }^{-}$ NO 175.0997, found 175.0996.

3-Formyl-2,3,4,5-tetrahydro-2-methyl-1H-3-benzazepine (21a), 2-Formyl-1,2,3,4-tetrahydro-1,1-dimethylisoquinoline (22), and $N$-(1-Methylethenyl)- $N$-(2-phenylethyl)formamide (23a). Following the general procedure, a boiling solution of $\mathbf{1 4 a}(100 \mathrm{mg}, 0.373 \mathrm{mmol})$ in toluene $(20 \mathrm{~mL})$ was treated with $\mathrm{Bu}_{3} \mathrm{SnH}(163 \mathrm{mg}, 0.559 \mathrm{mmol})$ and ACN $(18.2 \mathrm{mg}$, $0.0746 \mathrm{mmol})$ in toluene $(20 \mathrm{~mL})$. After workup, the crude material was chromatographed on silica gel (hexane/AcOEt, 6:1 $\rightarrow 3: 2)$. The first fraction gave $\mathbf{2 3 a}(5.5 \mathrm{mg}, 8 \%)$ as a colorless oil: IR $\left(\mathrm{CHCl}_{3}\right) v 1670,1645,1630 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H} \operatorname{NMR}(270 \mathrm{MHz}$, $\left.\mathrm{CDCl}_{3}\right) \delta 1.99(3 \mathrm{H}, \mathrm{s}), 2.83(2 \mathrm{H}, \mathrm{t}$ like, $J=7.9 \mathrm{~Hz}), 3.79(2 \mathrm{H}$, t like, $J=7.9 \mathrm{~Hz}), 4.61(1 \mathrm{H}, \mathrm{s}), 4.62(1 \mathrm{H}, \mathrm{s}), 7.19-7.32(5 \mathrm{H}$, $\mathrm{m}), 8.42(1 \mathrm{H}, \mathrm{s}) ;{ }^{13} \mathrm{C}$ NMR ( $67.8 \mathrm{MHz}, \mathrm{CDCl}_{3}$ ) $\delta 20.1,33.6,43.1$, 102.8, 126.5, 128.5, 128.8, 138.5, 141.4, 160.8. HRMS calcd for $\mathrm{C}_{12} \mathrm{H}_{15} \mathrm{NO}$ 189.1154, found 189.1153. The second fraction gave a mixture of 21 a and $22(50.8 \mathrm{mg}, 72 \%)$ in a ratio of 93:7: IR $\left(\mathrm{CHCl}_{3}\right) v 1660 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H} \operatorname{NMR}\left(270 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 0.93(93 / 200$ $\times 3 \mathrm{H}, \mathrm{d}, J=6.9$, for 21a), $1.04(93 / 200 \times 3 \mathrm{H}, \mathrm{d}, J=6.9 \mathrm{~Hz}$, for 21a), $1.75(7 / 100 \times 6 \mathrm{H}$, s, for 22), $2.72-3.00(93 / 100 \times 3 \mathrm{H}+$ 93/200 H, m, for 21a), 3.11-3.21 (93/100 H, m, for 21a), 3.25$3.35(93 / 200 \mathrm{H}, \mathrm{m}$, for 21a), $3.54-3.62(93 / 200 \mathrm{H}$, m, for 21a), $3.84(7 / 100 \times 2 \mathrm{H}, \mathrm{t}, ~ J=5.8 \mathrm{~Hz}$, for 22$), 3.97-4.03(93 / 200 \mathrm{H}$, m , for 21a), $4.50-4.57(93 / 200 \mathrm{H}, \mathrm{m}$, for 21a), 5.00-5.06 (93/ $200 \mathrm{H}, \mathrm{m}$, for 21a), $7.06-7.24(93 / 100 \times 4 \mathrm{H}+7 / 200 \times 4 \mathrm{H}, \mathrm{m}$, for 21a and 22), $8.08(93 / 200 \mathrm{H}$, s, for 21a), $8.17(93 / 200 \mathrm{H}$, s, for 21a), $8.63\left(7 / 100 \mathrm{H}\right.$, s, for 22); ${ }^{13} \mathrm{C}$ NMR for 21a ( 67.8 MHz , $\left.\mathrm{CDCl}_{3}\right) \delta 15.5,17.1,36.9,37.9,41.6,43.1,44.2,52.2,126.7,126.9$,
127.1, 129.2, 129.4, 130.6, 130.8, 136.7, 137.5, 139.5, 140.4, 161.7, 161.9. ${ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR spectra of 21a showed it to be a mixture of rotamers. HRMS calcd for $\mathrm{C}_{12} \mathrm{H}_{15} \mathrm{NO}$ 189.1154, found 189.1152.

Radical Cyclization of $\mathbf{1 4 b}$ at Room Temperature. To a solution of $\mathbf{1 4 b}(91 \mathrm{mg}, 0.323 \mathrm{mmol})$ and V-70 ( $50.0 \mathrm{mg}, 0.161$ $\mathrm{mmol})$ in toluene $(20 \mathrm{~mL})$ was added $\mathrm{Bu}_{3} \mathrm{SnH}(155 \mathrm{mg}, 0.532$ mmol ) at room temperature, and the mixture was stirred at the same temperature for 10 h . After workup, the crude material was chromatographed on silica gel (hexane/AcOEt, $4: 1 \rightarrow 3: 1$ ). The first fraction gave a mixture of $\mathbf{1 4 b}$ and $\mathbf{2 3 b}(31.1 \mathrm{mg})$ as a colorless oil. The second fraction gave 21b $(8.6 \mathrm{mg}, 13 \%)$ as a colorless oil.
$\boldsymbol{N}$-[2-(6-Bromobenzo[1,3]dioxol-5-yl)ethyl]- $\boldsymbol{N}$-(cyclopent-1-enyl)acrylamide (29). A mixture of 2 -(6-bromo-1,3-benzo-dioxol-5-yl)ethylamine ( $491 \mathrm{mg}, 2.01 \mathrm{mmol}$ ) and cyclopentanone ( $200 \mathrm{mg}, 2.38 \mathrm{mmol}$ ) in benzene $(10 \mathrm{~mL})$ was heated under reflux with azeotropic removal of water for 2 h . After cooling, $\mathrm{Et}_{3} \mathrm{~N}$ ( $407 \mathrm{mg}, 4.02 \mathrm{mmol}$ ) and acryloyl chloride ( $267 \mathrm{mg}, 2.95 \mathrm{mmol}$ ) were added to the reaction mixture, and the mixture was stirred at room temperature for 20 min . The reaction mixture was diluted with water and extracted with AcOEt. The organic layer was washed with brine, dried $\left(\mathrm{MgSO}_{4}\right)$, and concentrated. The residue was chromatographed on silica gel (hexane/AcOEt, 5:1) to give $29(467 \mathrm{mg}, 64 \%)$ as colorless crystals, $\mathrm{mp} 96.5-97.5^{\circ} \mathrm{C}$ (hexane); $\mathrm{IR}\left(\mathrm{CHCl}_{3}\right) v 1645 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H} \operatorname{NMR}\left(270 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta$ 1.98 ( 2 H , quint., $J=7.6 \mathrm{~Hz}$ ), 2.37-2.42 ( $4 \mathrm{H}, \mathrm{m}$ ), 2.90-2.95 (2 $\mathrm{H}, \mathrm{m}), 3.62-3.68(2 \mathrm{H}, \mathrm{m}), 5.43(1 \mathrm{H}, \mathrm{s}), 5.60(1 \mathrm{H}, \mathrm{dd}, J=9.9$, $2.3 \mathrm{~Hz}), 5.95(2 \mathrm{H}, \mathrm{s}), 6.36(1 \mathrm{H}, \mathrm{dd}, J=16.8,2.3 \mathrm{~Hz}), 6.51(1 \mathrm{H}$, $\mathrm{dd}, J=16.8,9.9 \mathrm{~Hz}), 6.78(1 \mathrm{H}, \mathrm{s}), 6.97(1 \mathrm{H}, \mathrm{s}) ;{ }^{13} \mathrm{C}$ NMR ( 67.8 $\left.\mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 22.4,30.4,32.8,34.4,45.9,101.6,110.7,112.6$, 114.4, 126.9, 129.0, 131.3, 142.2, 147.0, 147.4, 165.5. Anal. Calcd for $\mathrm{C}_{17} \mathrm{H}_{18} \mathrm{BrNO}_{3}$ : C, 56.06; H, 4.98; N, 3.85. Found: C, 56.28; H, 5.00; N, 3.79.
(3a $R^{*}, 14 \mathrm{bS}{ }^{*}$ )-1,2,3,8,9,14b-Hexahydro-4H-cyclopenta $[a]$ -[1,3]dioxolo[4,5-h]pyrrolo[2,1-b][3]benzazepin-6(5H)-one (30). Following the general procedure, a boiling solution of 29 ( $200 \mathrm{mg}, 0.549 \mathrm{mmol}$ ) in toluene ( 50 mL ) was treated with $\mathrm{Bu}_{3^{-}}$ $\mathrm{SnH}(238 \mathrm{mg}, 0.818 \mathrm{mmol})$ and $\mathrm{ACN}(13.4 \mathrm{mg}, 0.0548 \mathrm{mmol})$ in toluene ( 50 mL ). The solvent was evaporated off and the residue was taken up in $\mathrm{CH}_{3} \mathrm{CN}$. This mixture was washed with hexane and concentrated. The residue was chromatographed on silica gel (hexane/AcOEt, 1:2) to give $\mathbf{3 0}(50.2 \mathrm{mg}, 32 \%)$ as colorless crystals. A sample was recrystallized from hexane/AcOEt for X-ray crystallographic analysis (see Supporting Information): $\operatorname{mp} 176-179{ }^{\circ} \mathrm{C}$; IR $\left(\mathrm{CHCl}_{3}\right)$ v $1670 \mathrm{~cm}^{-1} ;{ }^{1} \mathrm{H}$ NMR $(270 \mathrm{MHz}$, $\left.\mathrm{CDCl}_{3}\right) \delta 1.50-2.29(10 \mathrm{H}, \mathrm{m}), 2.55-2.66(1 \mathrm{H}, \mathrm{m}), 2.98(1 \mathrm{H}$, $\mathrm{dd}, J=10.9,8.6 \mathrm{~Hz}), 3.07-3.24(2 \mathrm{H}, \mathrm{m}), 3.96-4.09(1 \mathrm{H}, \mathrm{m})$, $5.89(1 \mathrm{H}, \mathrm{d}, J=1.7 \mathrm{~Hz}), 5.90(1 \mathrm{H}, \mathrm{d}, J=1.7 \mathrm{~Hz}), 6.59(1 \mathrm{H}, \mathrm{s})$, $6.62(1 \mathrm{H}, \mathrm{s}) ;{ }^{13} \mathrm{C} \operatorname{NMR}\left(67.8 \mathrm{MHz}, \mathrm{CDCl}_{3}\right) \delta 24.0,29.7,30.2$, $33.7,35.4,38.5,38.6,57.8,69.6,100.9,110.7,111.0,129.8,133.0$, 146.2, 146.5, 175.5. Anal. Calcd for $\mathrm{C}_{17} \mathrm{H}_{19} \mathrm{NO}_{3}: \mathrm{C}, 71.56 ; \mathrm{H}, 6.71$; N, 4.91. Found: C, 71.37 ; H, 6.62; N, 4.89. Several byproducts were also formed, but their structures were unknown at the moment.

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Supporting Information Available: Experimental procedure for 7, 8, 9, 10, 11b-d, 12a-d, 13b-d, 14a-d, 18b$\mathbf{d}, \mathbf{1 9 c}, \mathbf{d}, 20 b-\mathbf{d}, 21 b-\mathbf{d}$, and $23 \mathrm{~b}, \mathbf{c} ;{ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR spectra for $6,9,11 \mathrm{c}, \mathrm{d}, 13 \mathrm{c}, \mathrm{d}, 14 \mathrm{~b}, 17,18 \mathrm{a}, \mathrm{b}, 18 \mathrm{~d}, 19 \mathrm{a}, 19 \mathrm{~d}, 20 \mathrm{a}-\mathrm{d}$, 21a-d, 22, and 23a-c ( ${ }^{1} \mathrm{H}$ NMR spectrum only for 23b); and X-ray crystallographic data for 30 in CIF format. This material is available free of charge via the Internet at http://pubs.acs.org.
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[^0]:    $\dagger$ Kanazawa University.
    \# Kinki University.
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